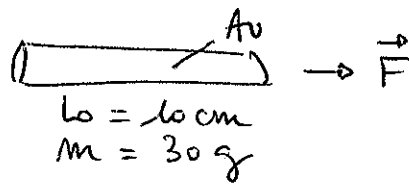


1. Allonger de l'or



$$1.1. V = \frac{m A_0}{\rho_{Au}} = \frac{m A_0}{d_{Au} \rho_{Au}} = \frac{30 \cdot 10^{-3}}{19,25 \times 10^3} = 1,56 \cdot 10^{-6} \text{ m}^3$$

$$1.2. V = S_0 \times L_0 \Rightarrow S_0 = \frac{V}{L_0} = \frac{1,56 \cdot 10^{-6}}{10 \cdot 10^{-2}} = 1,56 \cdot 10^{-5} \text{ m}^2$$

$$1.3. \frac{F}{S} = E \frac{L-L_0}{L_0} \Rightarrow F = ES \frac{L-L_0}{L_0}$$

$$= 78 \cdot 10^9 \times 1,56 \cdot 10^{-5} \times \frac{0,1}{100}$$

$$= 1,217 \cdot 10^3 \text{ N}$$

$$1.4. \frac{D_0 - D}{D_0} = \nu \frac{L-L_0}{L_0} \Rightarrow D = D_0 \left(1 - \nu \frac{L-L_0}{L_0} \right)$$

$$\text{or } S_0 = \pi \left(\frac{D_0}{2} \right)^2 \Rightarrow D_0 = 2 \sqrt{\frac{S_0}{\pi}} = 2 \times \sqrt{\frac{1,56 \cdot 10^{-5}}{\pi}} = 4,457 \cdot 10^{-3} \text{ m}$$

$$\text{donc } D = 2 \sqrt{\frac{S_0}{\pi}} \left(1 - \nu \frac{L-L_0}{L_0} \right)$$

$$= 2 \times \sqrt{\frac{1,56 \cdot 10^{-5}}{\pi}} \left(1 - 0,42 \times \frac{0,1}{100} \right) = 4,455 \cdot 10^{-3} \text{ m}$$

$$1.5. \left(\begin{array}{c} \omega \\ L_0 = 10 \text{ cm} \\ m = 30 \text{ g} \end{array} \right) \rightarrow \vec{F}$$

$$\frac{L-L_0}{L_0} = \frac{F}{ES} = \frac{F}{E} \frac{d_{\omega} \rho_{\omega} L_0}{m \omega} = 0,03\% /$$

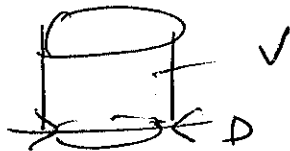
$$S_0 = \frac{V}{L_0} = \frac{m \omega}{d_{\omega} \rho_{\omega} L_0} = \frac{30 \cdot 10^{-3}}{19,25 \times 10^3 \times 10 \cdot 10^{-2}} = 3,359 \cdot 10^{-5} \text{ m}^2$$

$$\frac{D_0 - D}{D_0} = \nu \frac{L-L_0}{L_0} = \nu \frac{F}{ES} \quad D_0 = 2 \sqrt{\frac{S_0}{\pi}} = 6,540 \cdot 10^{-3} \text{ m}$$

$$D = D_0 \left(1 - \nu \frac{F}{ES} \right) = 2 \sqrt{\frac{S_0}{\pi}} \left(1 - \nu \frac{F}{ES} \right) = 2 \times \sqrt{\frac{3,359 \cdot 10^{-5}}{\pi}} \left(1 - 0,42 \frac{1,217 \cdot 10^3}{124 \cdot 10^9 \times 3,359 \cdot 10^{-5}} \right)$$

$$= 6,539 \cdot 10^{-3} \text{ m}$$

2. Magique ?



$$2.1 \quad m_{eau} = \rho_{eau} V_{eau} = 1 \times 250 \cdot 10^{-3} = 250 \cdot 10^{-3} \text{ kg}$$

$$2.2 \quad F_{eau} = m_{eau} \times g = 250 \cdot 10^{-3} \times 9,8 = 2,45 \text{ N}$$

$$2.3 \quad P_{eau} = \frac{F_{eau}}{S} = \frac{F_{eau}}{\pi \left(\frac{D}{2}\right)^2} = \frac{2,45}{\pi \left(\frac{6,8 \cdot 10^{-2}}{2}\right)^2} = 675 \text{ Pa}$$

$$2.4 \quad F_{atm} = P_{atm} S = 1,013 \cdot 10^5 \times \pi \left(\frac{6,8 \cdot 10^{-2}}{2}\right)^2 = 378 \text{ N} < F_{eau}$$

3. Composition de l'air.

$$3.1 \quad x_{O_2} = 0,22 \quad x_{N_2} = 0,78$$

$$M_{air} = x_{O_2} M_{O_2} + x_{N_2} M_{N_2} = 0,22 \times 2 \times 16 + 0,78 \times 2 \times 14 = 28,88 \text{ g} \cdot \text{mol}^{-1}$$

$$3.2 \quad P_{O_2} = x_{O_2} P_{atm} = 0,22 \times 1,013 \cdot 10^5 = 2,2 \cdot 10^4 \text{ Pa}$$

$$P_{N_2} = x_{N_2} P_{atm} = 0,78 \times 1,013 \cdot 10^5 = 7,9 \cdot 10^4 \text{ Pa}$$

$$3.3 \quad \text{Etat 1} : \quad \text{Etat 2} :$$

$$P_1 = P_{atm}$$

$$V_1 = 1 \text{ L}$$

$$T_1 = 20^\circ \text{C}$$

$$P_2 = P_{atm}$$

$$V_2 = ?$$

$$T_2 = 37^\circ \text{C}$$

$$n_1 = \frac{P_1 V_1}{RT_1} = \frac{P_2 V_2}{RT_2} \Rightarrow V_2 = V_1 \frac{T_2}{T_1}$$

$$V_2 = 1 \times \frac{37 + 273}{20 + 273} = 1,058 \text{ L}$$

$$3.4 \quad P_{atm} = \sum_i P_i' \Rightarrow P_{N_2}' = P_{atm} - (P_{O_2}' + P_{H_2O}' + P_{CO_2}') = 75819 \text{ Pa}$$

$$3.5 \quad M'_{air} = \sum_i x_i' M_i = \sum_i \frac{P_i'}{P_{atm}} M_i = 28,77 \text{ g} \cdot \text{mol}^{-1}$$

4. Jeux de température

$$4.1 \quad L(T_1) = L(T_0) (1 + \lambda (T_1 - T_0))$$

$$L(0^\circ \text{C}) = 10^{-5} (1 - 20 \times 12,6 \cdot 10^{-6}) = 0,99976 \cdot 10^{-5} \text{ m}$$

$$40,26 \times 0,99976 \cdot 10^{-2} = 40,25 \text{ mm}$$

4. Jeux de température

4.1. $L(T_1) = L(T_0) (1 + \lambda(T_1 - T_0))$

$L(0^\circ\text{C}) = 10^{-5} (1 - 20 \times 12 \cdot 10^{-6}) = 0,99976 \cdot 10^{-5} \text{ m}$

$\Rightarrow 40,26 \cdot 10^{-3} \times 0,99976 \cdot 10^{-5} = 40,25 \text{ mm}$

4.2 $\bar{a} \text{ } 0^\circ\text{C}$ $\sqrt{V_{(0^\circ\text{C})}^{\text{liq}}} = 1L$ $\sqrt{V_{(0^\circ\text{C})}^{\text{sol}}} = 1L$

$\bar{a} \text{ } 95^\circ\text{C}$ $\sqrt{V_{(95^\circ\text{C})}^{\text{liq}}} = \sqrt{V_{(0^\circ\text{C})}^{\text{liq}}} (1 + \alpha \times 95) = 1 \sqrt{1 + 2,1 \cdot 10^{-4} \times 95} = 1,01995L$

$\sqrt{V_{(95^\circ\text{C})}^{\text{sol}}} = \sqrt{V_{(0^\circ\text{C})}^{\text{sol}}} (1 + 3\lambda \times 95) = 1 \times (1 + 3 \times 8 \cdot 10^{-6} \times 95) = 1,00228L$

$\Delta V = \sqrt{V_{(95^\circ\text{C})}^{\text{liq}}} - \sqrt{V_{(95^\circ\text{C})}^{\text{sol}}} = 1,767 \cdot 10^{-2} L$

4.3 Etat 1 Etat 2

$T_1 = 14^\circ\text{C}$

$P_1 = 3 \text{ bar}$

$n_1 = \frac{P_1 V_1}{RT_1}$

$T_2 = 20^\circ\text{C}$

$P_2 = ?$

$n_2 = \frac{P_2 V_2}{RT_2}$

$P_2 = P_1 \frac{T_2}{T_1} = 3 \times \frac{20 + 273}{37 + 273}$

$P_2 = 2,13 \text{ bar}$

4.4. $L(T_1) = L(T_0) (1 + \lambda(T_1 - T_0))$

$\lambda = \left(\frac{L(T_1)}{L(T_0)} - 1 \right) \times \frac{1}{T_1 - T_0}$

$= \left(\frac{300}{299,946} - 1 \right) \times \frac{1}{15}$

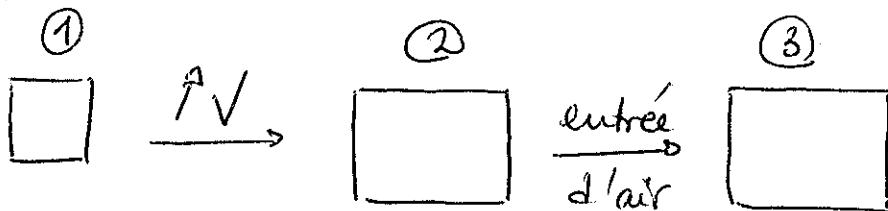
$= 12 \cdot 10^{-6} \text{ } ^\circ\text{C}^{-1} = \lambda_{\text{Acier}}$

$L(T_1 = 15^\circ\text{C}) = 300 \text{ m}$

$L(T_0 = 0^\circ\text{C}) = 300 - 5,4 \cdot 10^{-2}$

$= 299,946 \text{ m}$

5. Respiration et ventilation assistée



$P_1 = P_{\text{atm}}$

$V_1 = 2,5L$

$T_1 = T_{\text{air}}$

n_1

P_2

$V_2 = 3L$

$= T_2$

$n_2 = n_1$

③

$P_3 = P_{\text{atm}}$

$V_3 = V_2 = 3L$

$= T_3 = 37^\circ\text{C}$

$n_3 > n_1$

$$5.1. \quad m_1 = \frac{P_1 V_1}{RT_1} = \frac{P_2 V_2}{RT_2} \quad P_2 = P_1 \frac{V_1}{V_2} = 760 \times 133,3 \times$$

$$P_2 = P_1 \frac{V_1}{V_2} =$$

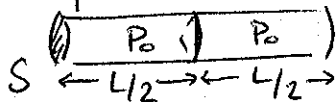
$$5.2. \quad m = m_3 - m_1 = \frac{P_3 V_3}{RT_3} - \frac{P_1 V_1}{RT_1} = \frac{P_{atm}}{RT_{avr}} (V_3 - V_1)$$

$$= \frac{760 \times}{8,31 \times (37 + 273)} (0,5 \cdot 10^{-3}) = 0,02 \text{ mol}$$

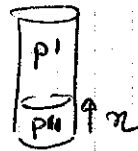
$$5.3. \quad F = F_{int} - F_{ext} = (P_V - P_{atm}) S_{piston}$$

$$= (763 - 760) \times 133,3 \times 200 = 80 \cdot 10^3 \text{ N}$$

6. Un piston dans un tube



→



6.1. Equilibre

$$Mg + P'S - P''S = 0$$

6.2.

$$m_0 = \frac{P_0 V_0}{RT_0} = m' = \frac{P' V'}{RT'} = m'' = \frac{P'' V''}{RT''}$$

$$P' = \frac{P_0 V_0}{V'} \quad P'' = \frac{P_0 V_0}{V''}$$

$$P' = P_0 \frac{L}{2(L-x)} \quad P'' = P_0 \frac{L}{2x}$$

$$\left. \begin{array}{l} V_0 = \frac{S L}{2} \\ V' = S(L-x) \\ V'' = Sx \end{array} \right\}$$

6.3.

$$Mg + P'S - P''S = 0 = Mg + P_0 \frac{L}{2(L-x)} - P_0 \frac{L}{2x} = 0$$

$$(*) \Leftrightarrow 2a^2 - 2(1+a)x + a = 0 \quad \text{avec } a = \frac{SP_0}{Mg}$$

6.4.

$$a = \frac{SP_0}{Mg} = \frac{1 \cdot 10^{-4} \times 1,013 \cdot 10^5}{0,5 \times 10} = 2$$

$$(*) \Leftrightarrow x^2 - 3x + 1 = 0 \quad \Delta = 5 \quad \begin{array}{l} x_1 = 0,382 \text{ m} \\ x_2 = 2,618 \text{ m} \end{array} > L$$

$$x = 0,382 \text{ m}$$

$$P' = P_0 \frac{L}{2(L-x)} = 1 \cdot \frac{1}{2(1-0,382)} = 0,81 \text{ bar}$$

$$P'' = P_0 \frac{L}{2x} = 1 \cdot \frac{1}{2 \times 0,382} = 1,31 \text{ bar}$$